

## Chemistry 100 Practice Final Exam Instructions

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1 Chem 100 Practice Final Exam 1. Which one of the following numbers contains 4 significant figures? a. 6.43 b. 39.4 c. 0.04840 d. 0.0257 e. none of the above 2. In which of the following pairs of numbers does each member of the pair contain the same number of significant figures? 3. A balance has an accuracy of  $\pm 0.01$  grams.

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Chemistry 100 Practice Final Exam Instructions: Do not begin the exam until you have been instructed to do so. You have 120 minutes to complete this exam. There are 50 multiple choice questions. You must use a number 2 pencil. You may use a scientific calculator. Make sure that you have written your name legibly on the scantron form.

Chemistry 100 Practice Final Exam Instructions

Chemistry 100 Final Exam (ALL SECTIONS) Multiple Choice Questions (12 points, 2 each): answer the following multiple choice questions by circling the best response in each case. 1. The Cl-C-Cl bond angle in the COCl<sub>2</sub> molecule is approximately: 92o. 104.5o.

Chemistry 100 Final Exam (ALL SECTIONS)

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Chemistry 100 Practice Final Exam Instructions

Chemistry 100 Final Exam (ALL SECTIONS) April 15, 2009. M. Aquino; G. Orlova; B. MacLean. Name: \_\_\_\_\_ ID: \_\_\_\_\_. This exam has a three-hour (3 hour) time limit. Read over the entire exam before beginning. Do those that you find easiest in the questions beginning.

Chemistry 100 Final Exam (ALL SECTIONS)

Chem. 100 Exam II (Practice Test) ATOMIC STRUCTURE: The following information applies to subatomic particles. If the correct answer is Proton, mark [A.] neutron, mark [B.] electron, mark [C.] If the answer is correct for more than one, mark all correct answers. 1. The mass of the particle doesn't weigh enough to worry about. 2.

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Practice Exams. Practice Exam 1. Answers to PE1. Practice Exam 2. Answers to PE2. Practice Exam 3. Answers to PE3 pgs1-5. CH141 Practice Exam III Key B. Practice Final Exam Problems. PF answers pg1-6. CH141 Practice Final Key II (pages 6-12) CH141 Exam I 2016 with Answers. CH141 Exam II 2016 with Answers. CH141 Exam III 2016 with Answers ...

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Chemistry 30 Practice Final Exam page 7. H H. H H. H H. H H. volume of base (mL.) more space to answer rest of question on next page. Title: Chemistry 30 Practice Final Exam Author: Rob Schultz Last modified by: Rob Schultz Created Date: 12/29/2008 11:33:00 PM Company: Home

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Chemistry 100 Practice Final Exam Instructions

Chemistry 100 Exam I Roadmap to Chemistry Mass Calculations Grams of element Molar mass of element (g/mol) Moles of element SYRJJDGURQV number of atoms or ions Atoms or Ions Atomic mass (amu / atom) Amu of element Grams of compound A Molar mass of compound A Moles of compound A SYRJJDGURQV number of molecules Molecules Formula mass (amu / molecule)

Instructor: Nicole Stevens 1 of 8 Winter 2008 Chemistry ...

Chemistry 11 1. Conserving significant figures. 0.0250 m is equivalent to: A. 0.25 cm B. 2.5 cm C. 2.50 cm D. 25.0 cm E. none of the above 2. In which of the following sets of quantities

Chemistry 11 Final Exam - Mrs Dildy

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Organic Chemistry 2 Final Exam Review Multiple Choice Test ...

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